

Charge Exchange Processes for Highly
Charged Ion - Atom, Molecule
Collisions

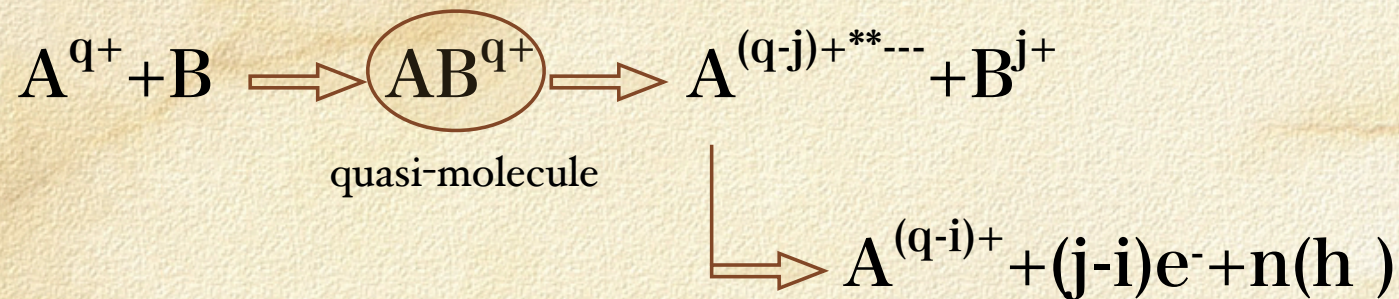
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Charge Exchange Processes

- Highly charged ions (HCI's) exist as an impurity ion in high temperature plasma.
- Charge exchange cross sections of HCI's are very large ($\sim 10^{-14} \text{cm}^2$)
- Charge Exchange Processes of HCI's are very important for understanding the edge plasma behavior in thermonuclear plasma
- Charge Exchange Spectroscopy (CXS) is one of the most important plasma diagnostic in LHD

Collision Processes



A^{q+} : Highly Charged Ion, B: Target Atom, e^{-} : Ejected Electron,

i -electron capture after j -electron transfer

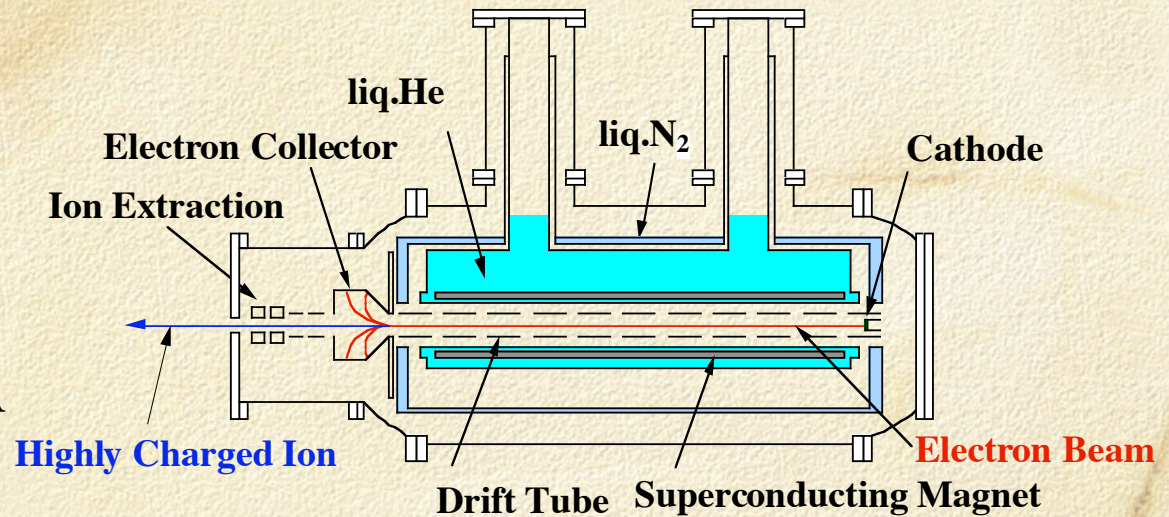
This collision process and decay process are greatly concerned with the cooling of the plasma. We paid attention to the following thing, and did research by using the superconductivity Electron Beam Ion Source (EBIS).

- How is an electron transferred?
- How large is transfer cross section?
- How is the charge dependence of the transfer cross section?

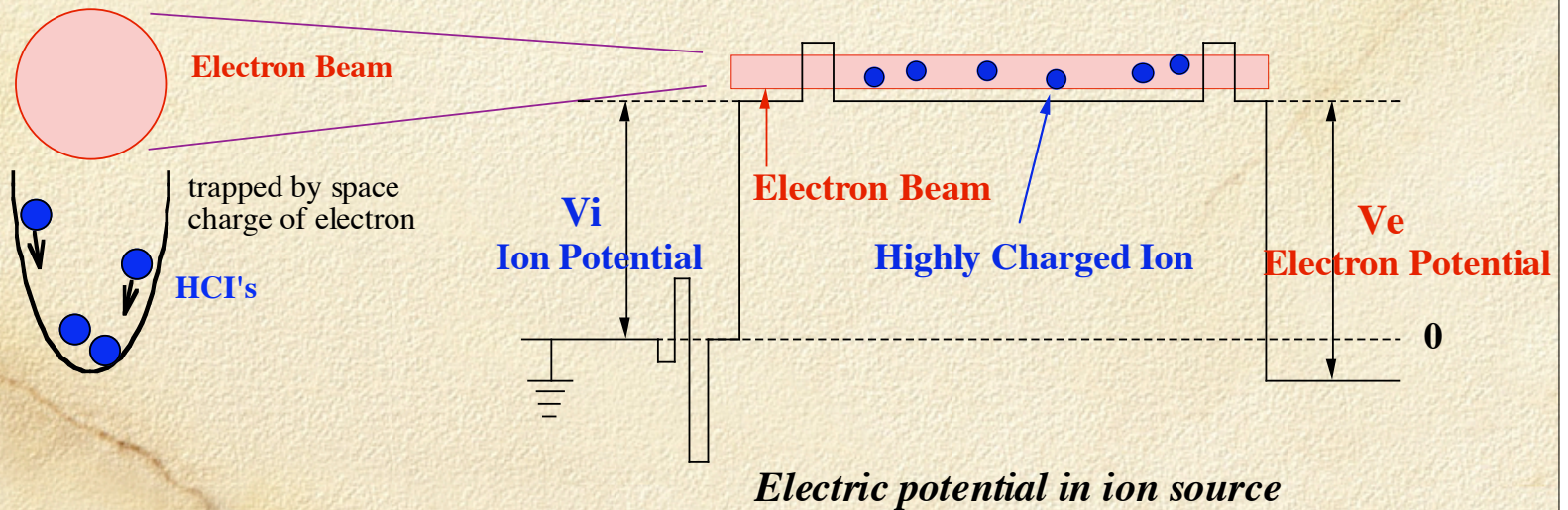
Highly Charged Ion Source

EBIS

Electron energy : $\sim 5\text{keV}$
Ion energy : 1.5keV
Magnetic field : 2T
Electron current : $\sim 10\text{mA}$

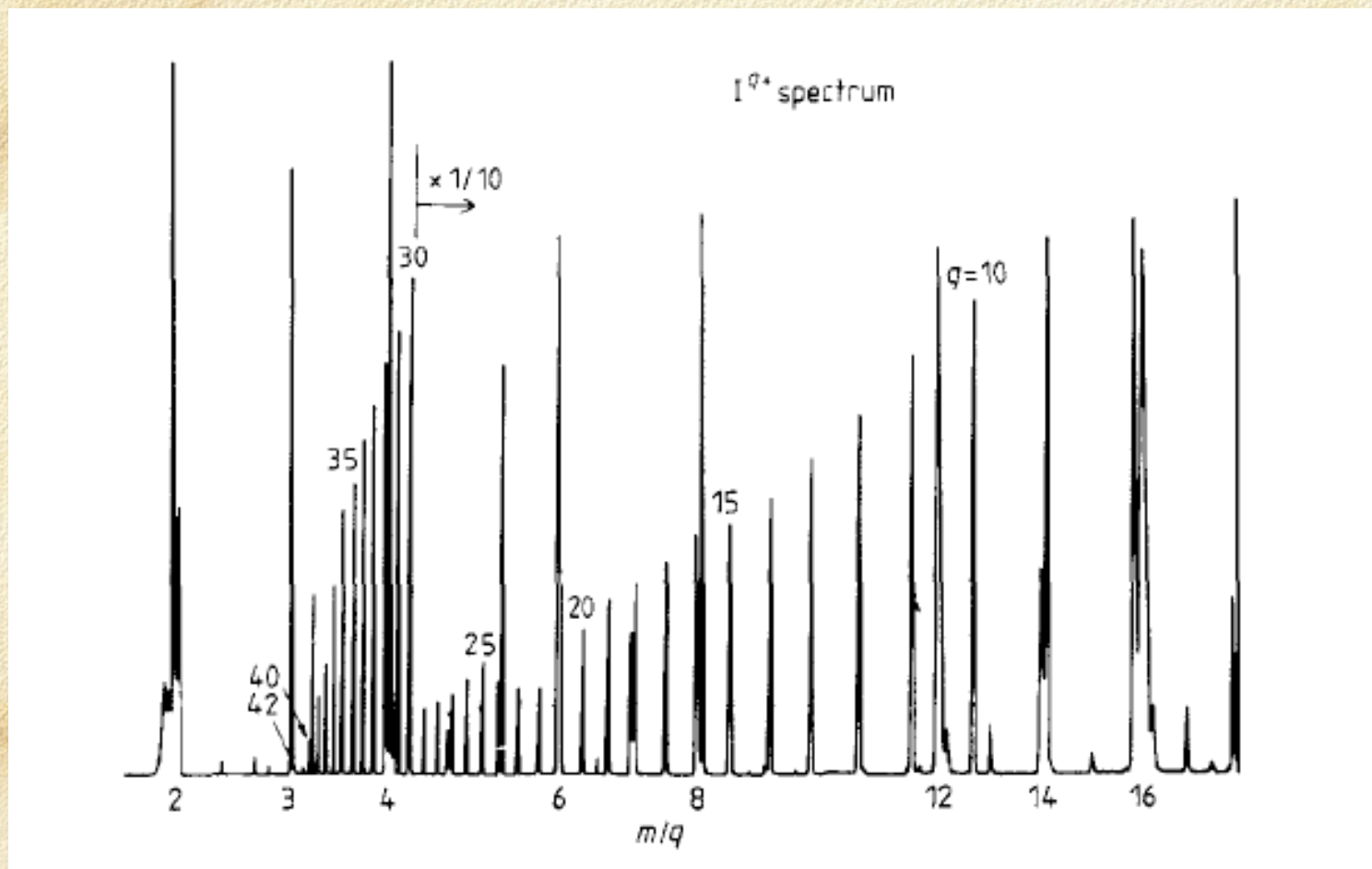


Electron Beam Ion Source (EBIS)



Mass Spectrum of EBIS

$^{127}_{53}\text{I}^q$

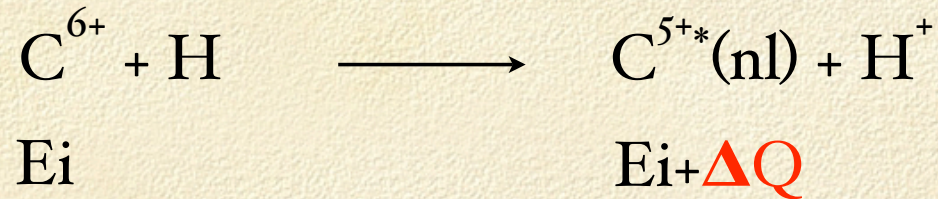


Experiments and Results

i. Energy Gain Spectroscopy \longrightarrow Energy level of electron transfer

We measure the kinematic energy $E_i + \Delta Q$ (eV) by energy analysis of the scattered ion.

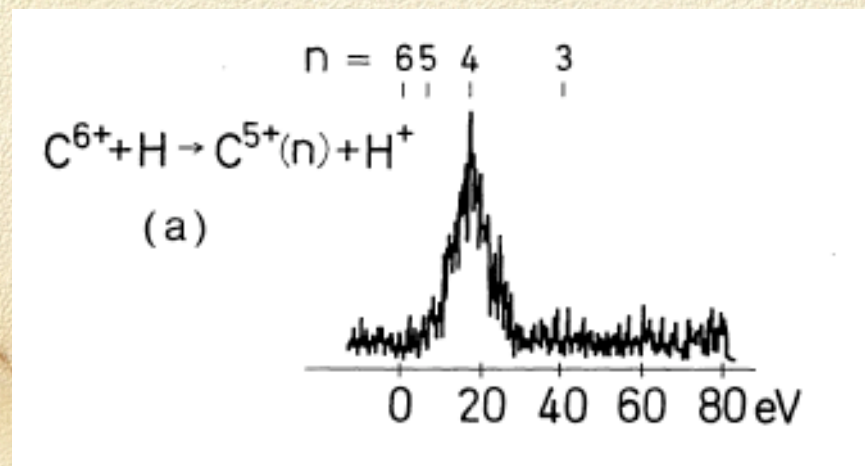
e.g. one-electron transfer processes



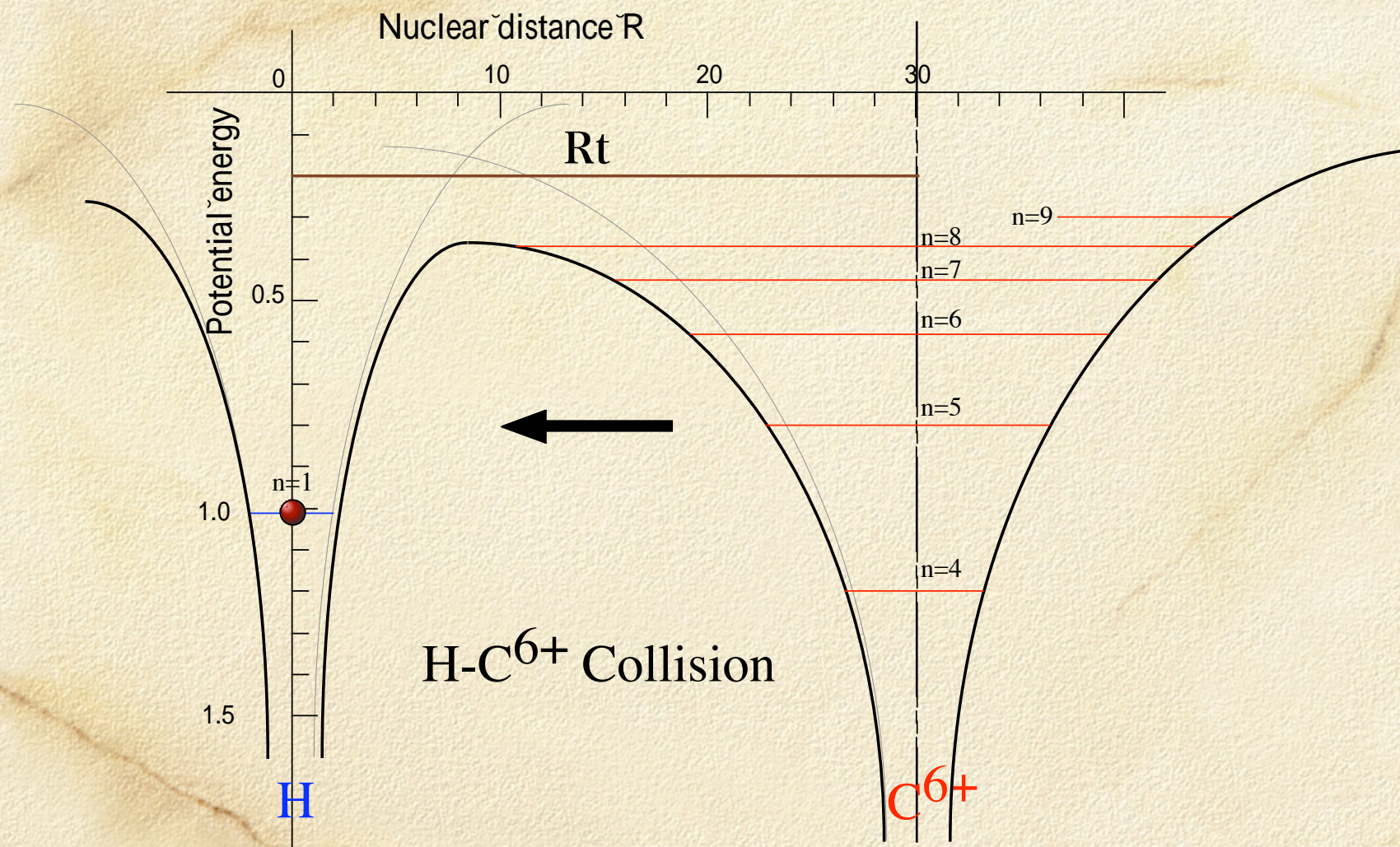
E_i : incidence energy

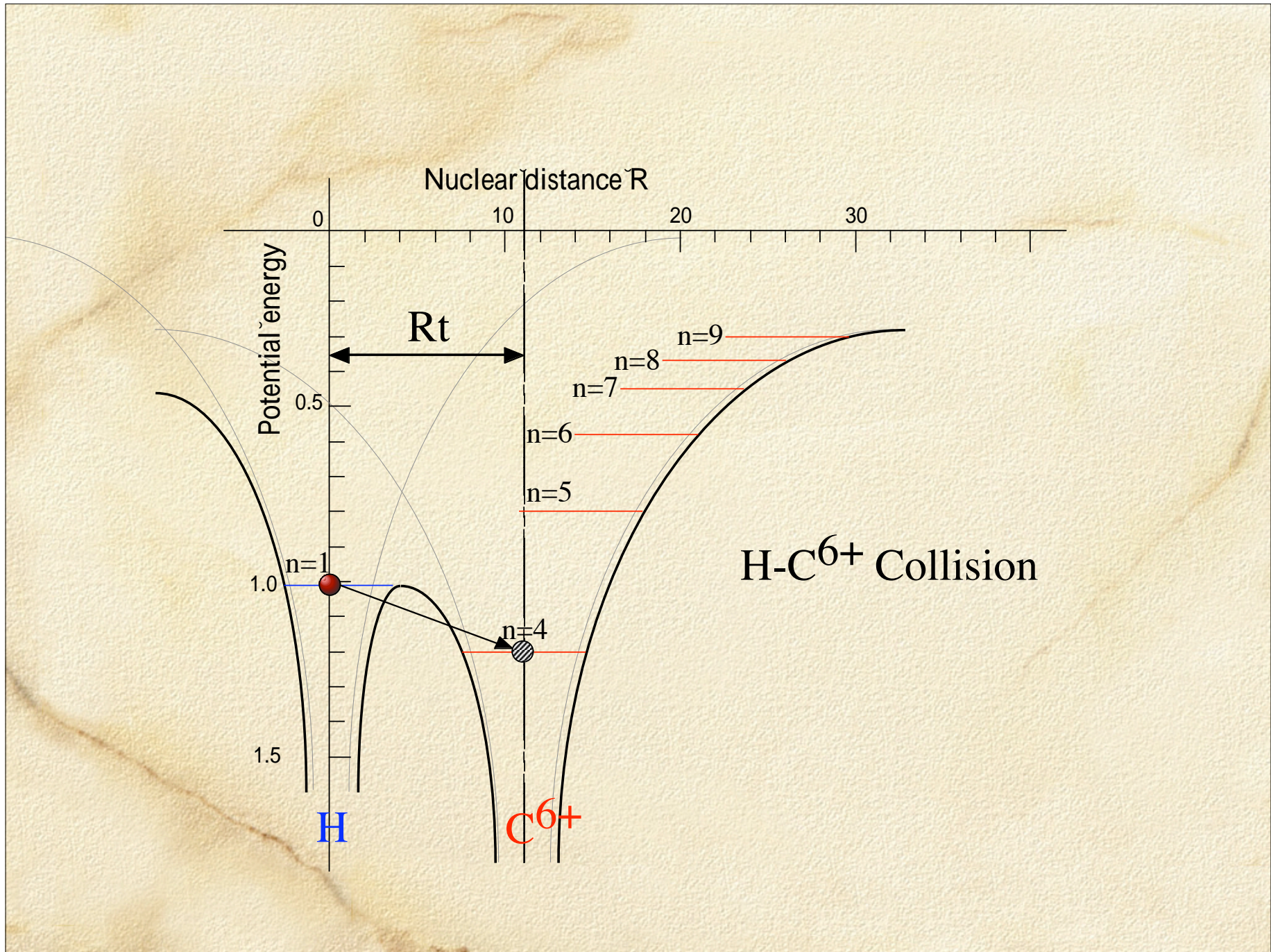
ΔQ : Energy Gain

n : principal quantum number



Classical over barrier model





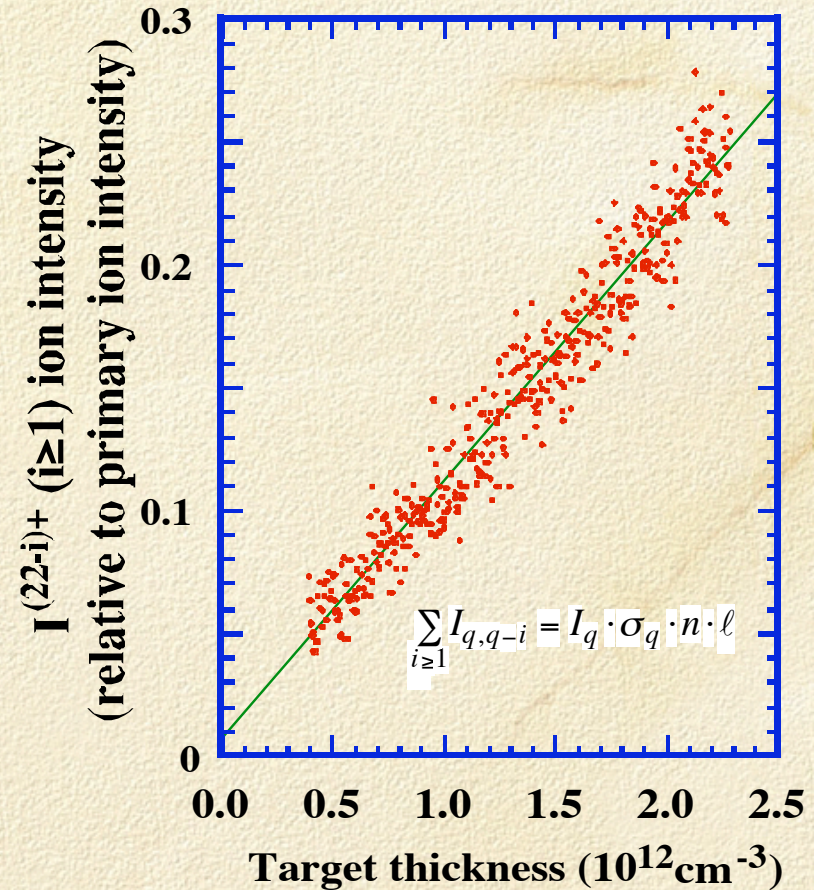
Experiments and Results

2. Initial Growth-Rate Method



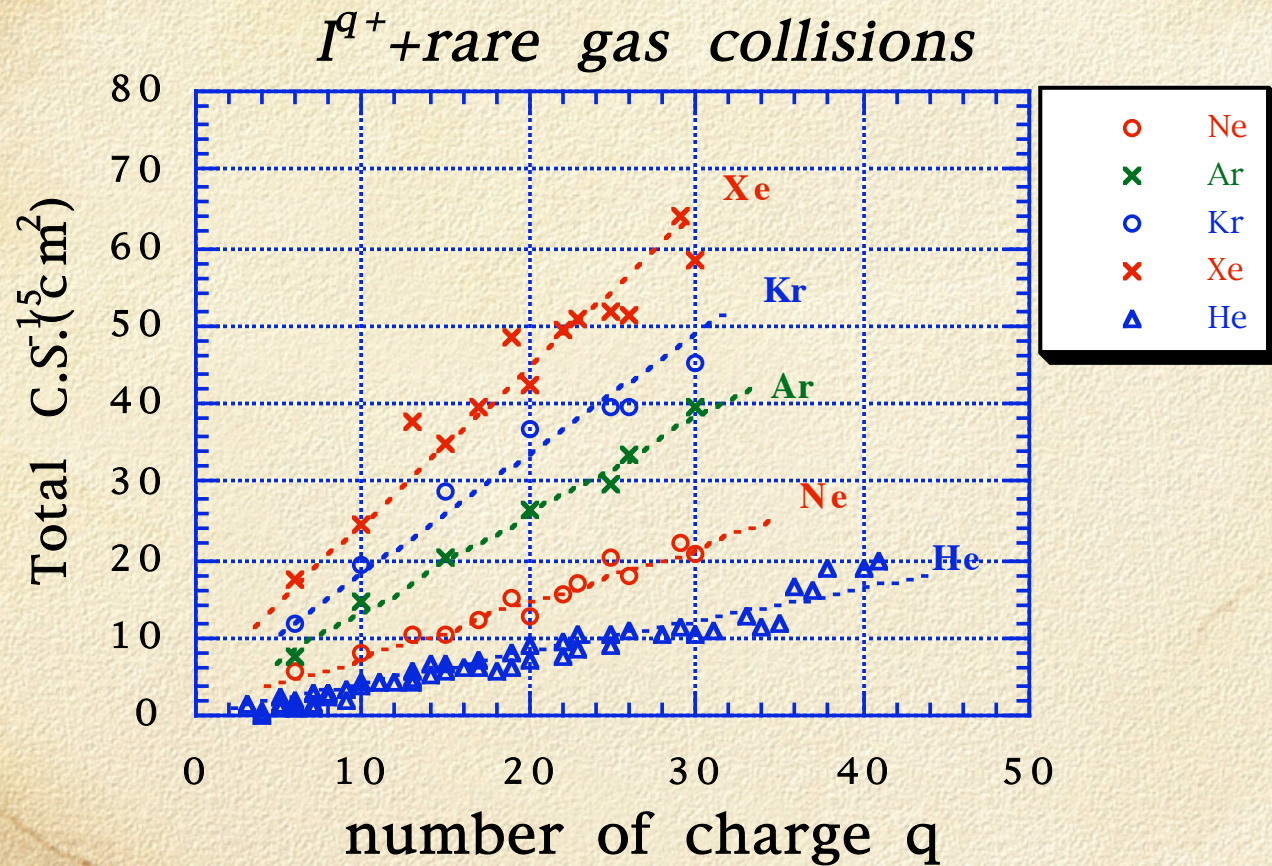
Electron transfer cross section

We can decide the total electron transfer cross sections from the gradient of that function of the scattered ion signal strength vs target density.

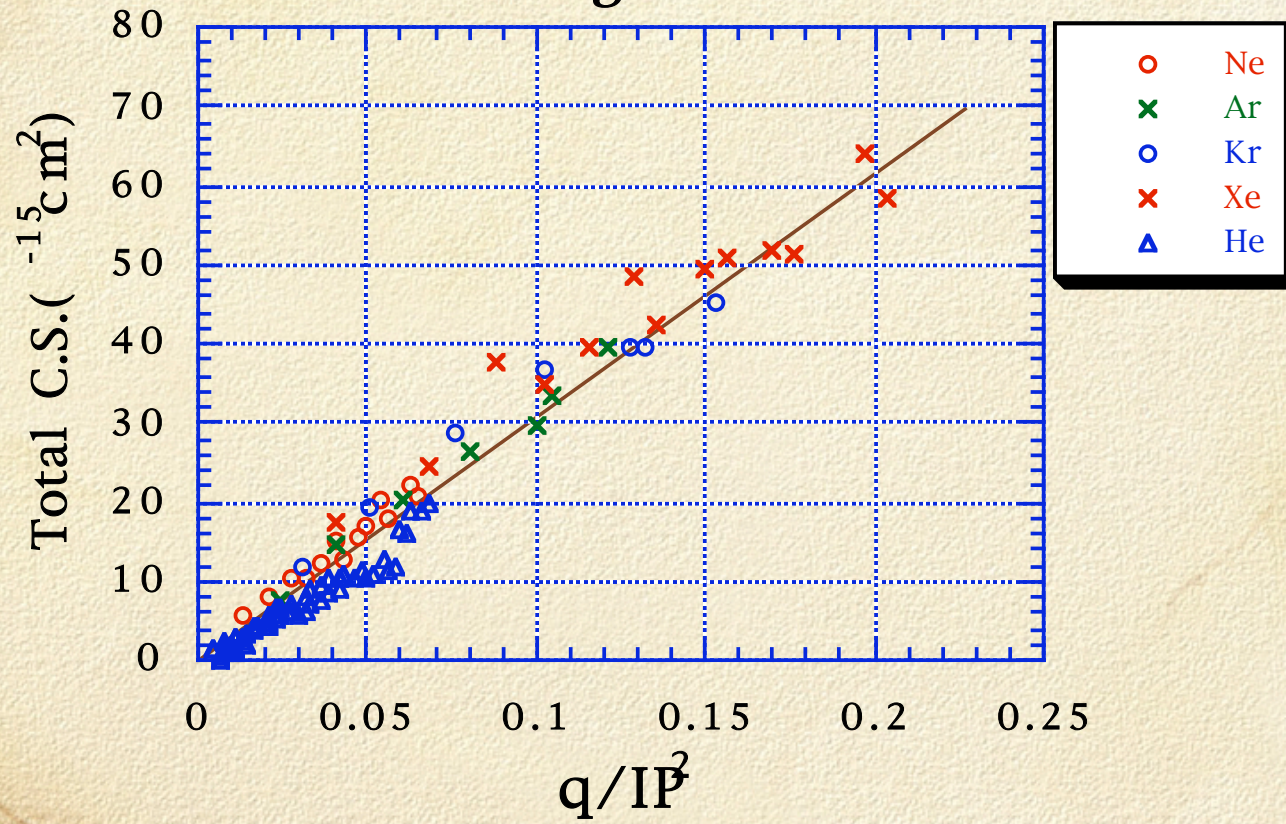


Experimental results

Charge Exchange Cross Section
(Electron Transfer)



I^{q+} + rare gas collisions



Scaling law for cross section

transfer condition

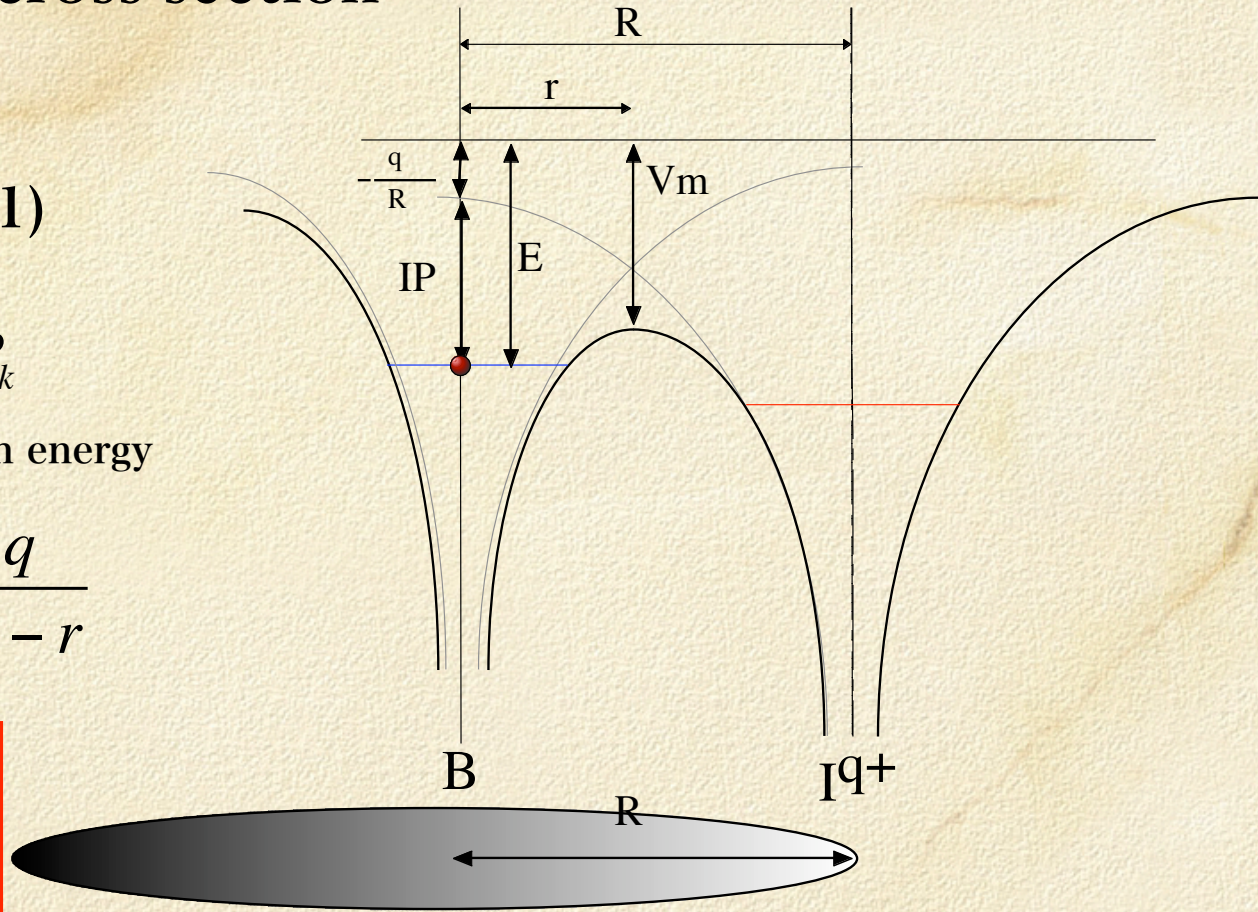
$$E = V_m \quad (1)$$

$$E = -\frac{q}{R} - IP_k$$

IP: ionization energy

$$V_m = -\frac{k}{r} - \frac{q}{R-r}$$

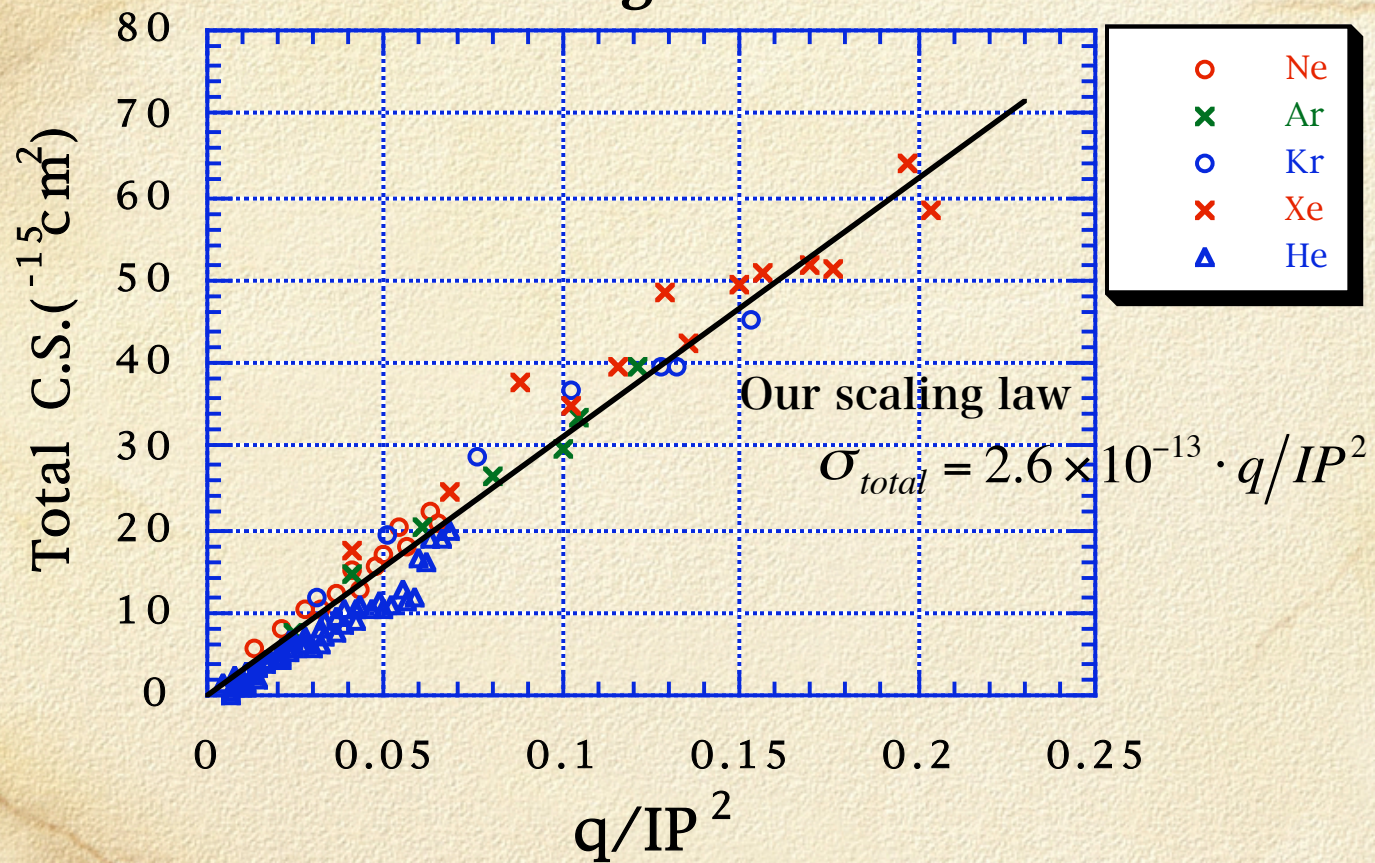
$$R = 2 \cdot \sqrt{q} \cdot \frac{1}{IP}$$



$$\sigma_{total} = \pi R^2 = 4\pi \cdot q / IP^2 = 2.6 \times 10^{-13} \cdot q / IP^2$$

(atomic unit) (cm²)

I^{q+} + rare gas collisions

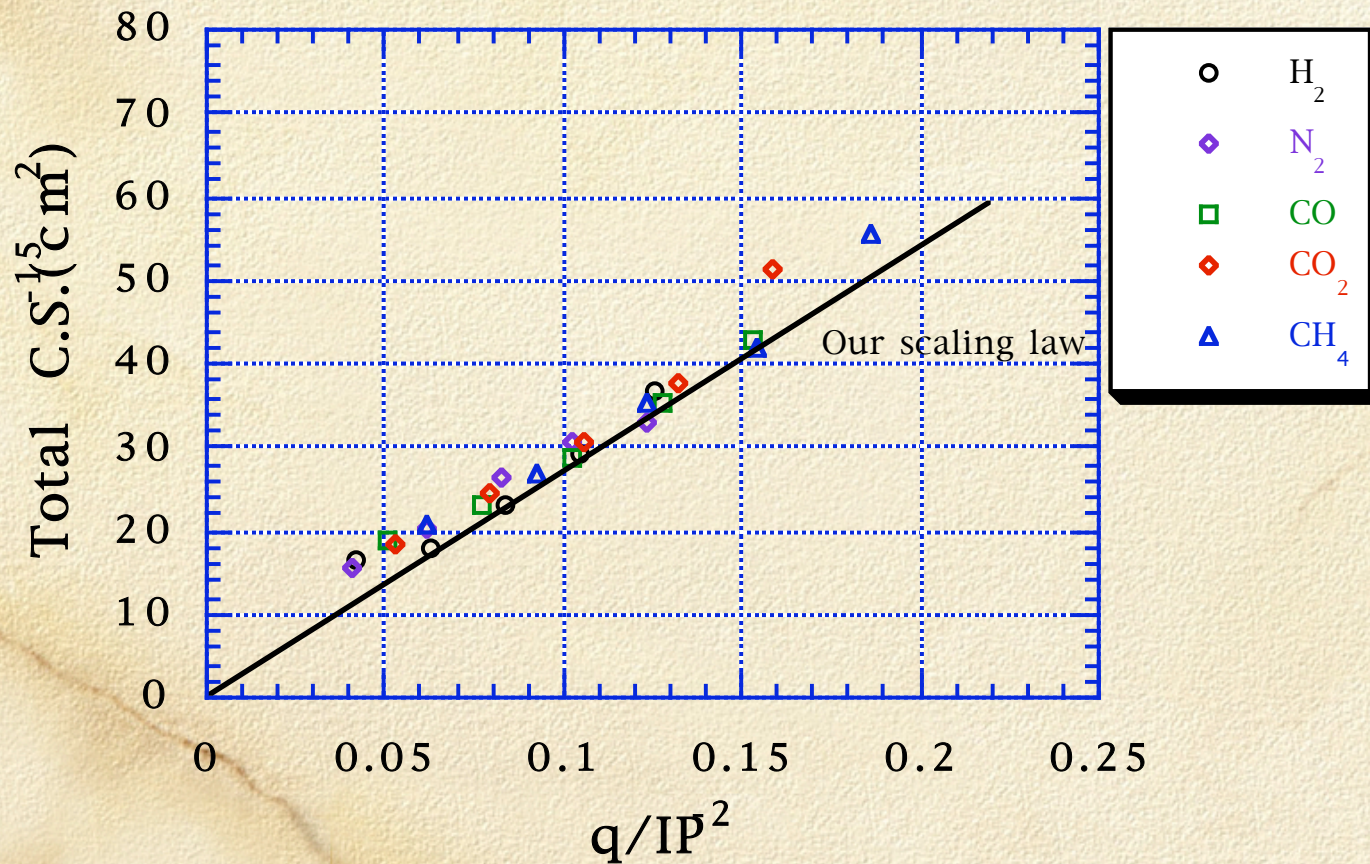


Experimental results molecule targets

IP(H₂) = 15.4 eV
IP(N₂) = 15.6 eV
IP(CO) = 14.0 eV
IP(CO₂) = 13.8 eV
IP(CH₄) = 12.6 eV

from Handbook of
Chemistry and Physics

$I^{q+} + \text{mol. collisions}$



Summary

- The detail of electron transfer process by HCl's-Atom collision has been made clear.
- The absolute electron transfer cross sections were determined.
- We proposed the scaling law of the electron transfer cross section.
- We have found that this scaling law can also reproduce the experimental data for molecular targets.
- Most particles are in the excited states in the plasma. The charge exchange cross sections of excited targets-HCl's collision are very large.

$$\text{e.g. } IP\{H^*(2s)\} : IP\{H(1s)\} = 3.4\text{eV} : 13.6\text{eV}$$

$$\sigma_{total} H(2s) / \sigma_{total} H(1s) = 16$$

$$\sigma_{total} = 2.6 \times 10^{-13} \cdot q / IP^2$$

scaling law

Thank you for your attentions.

감사합니다.

ありがとうございました。