

NSTX-U is sponsored by the U.S. Department of Energy Office of Science Fusion Energy Sciences

### Liquid Metals: Near and Long-term Plans

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NSTX-U Program Advisory Committee Meeting Princeton Plasma Physics Laboratory, Princeton, NJ January 9<sup>th</sup>, 2018







### Outline

- 1. Why evaluate liquid metals as a plasma-facing component (PFC)?
  - Liquid PFCs resolve several problems plaguing solid PFCs
  - Liquid PFCs demonstrated to enhance confinement and can absorb more heat flux than do solid PFCs
- 2. Why study liquid metals now and why in NSTX-U?
  - PMI Community workshop highlighted liquid PFCs in numerous priority research directions and high-impact cross-cutting activities
  - Integrated scenario assessment of ST requires movement away from C



### Liquid metals present *intrinsic* advantages over solid PFCs in several areas

- Liquid metals provide a selfhealing/renewable plasma-facing material
  - Immune to thermo-mechanical stresses
  - Returns to equilibrium after perturbations
  - Replenishment eliminates net-reshaping by plasma bombardment
- Separates neutron damage effects from plasma-material interactions
- Eliminates long-time constants associated with solid-wall material transport and evolution<sup>a</sup>
- Demonstrated confinement effects and power handling capabilities

<sup>a</sup>Stangeby, JNM 2011

### Cracking after thermal shock loading



#### Wirtz, et al., JNM 2013



Coenen, et al., JNM 2013

Mo first-wall



M. Reinke, private comm.



## LTX demonstrated (again) confinement gains even with partial oxidation of Li coatings and Li pool

- Measured electron energy confinement exceeds neo-Alcator Linear Ohmic Confinement scaling ~3x (Boyle PRL 2017)
- Evaporated coatings monitored with MAPP diagnostic correlated with good performance to ~100 hrs (Lucia PhD 2015, Schmitt PoP 2015)
- Key question: will flowing liquids perform like coatings (with impurities)?





Passivated Li walls

× Liquid Li walls

D. Boyle, et al., *Phys. Rev. Lett.* **119** (2017) 015001.

J. Schmitt, et al., Phys. Plasmas 22 (2015) 056112.

# Liquid metal PFCs can absorb more heat flux than do leading tungsten technologies

- Actively cooled tungsten expected to survive 5-15 MW m<sup>-2</sup> steady-state
- Fast-flow systems advect power away from heating zones
  - Limiting temperature, heating size, and velocity determine limiting heat flux
  - Li, Ga, Sn all possible metals for use
- Slowly flowing liquid targets recently demonstrated in multiple configurations
  - Non-vaporizing, water-cooled tin demonstrated at 20 MW m<sup>-2</sup> in Magnum-PSI<sup>a</sup>
  - Vapor-shielded tin achieved self-regulated temperature up to 22 MW m<sup>-2</sup> in Magnum-PSI<sup>b</sup>

<sup>a</sup>Morgan, 2017 Nucl. Mater. Energy; <sup>b</sup>S. Van Eden, et al., Nature Comm. 8 (2017) 192.





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#### US PMI community workshop highlights the importance of developing tools that can extrapolate to a reactor

- Multiple priority research directions (PRDs) highlight need to advance concepts (including **liquids**):
  - PRD-A: limits on power and particle handling, including liquids
  - PRD-B: demonstrate dissipative divertor solutions, including **liquids**
  - PRD-C: develop solutions for main chamber components and tools for sustained operation
  - PRD-D: science of evolving materials, including liquids
  - PRD-E: understand impact of materials on core performance
- Multiple high-impact, cross-cutting activities also identified:
  - CC-1: enhance exploitation of existing machines for PMI studies
  - CC-3: understand the science of liquid surfaces and examine feasibility of liquid PFC solutions
     See also Reinke's talk

# NSTX & PPPL expertise, and need for integrated solutions motivates for liquid metal mission in NSTX-U

- Years of developing Li science...
  - Diagnostics: dual-band IR (ORNL), divertor imaging/spect. (LLNL), MAPP (U-Illinois)
  - Theory: liquid stability, quantum molecular dynamics (SBU) and DFT (PU), wholedevice material evolution modeling
  - Supporting surface science laboratories
- ...and technologies
  - Lithium experimental methods: LITER, U-
    - LITER, granule & aerosol injection, Liquid Lithium Divertor
  - Large area tray limiters, LTX shell system, etc.
- NSTX-U can easily create  $q_{\perp} > 20$  MW m<sup>-2</sup>
  - Full heating power: P/R~20

ar



#### NSTX Liquid Li Divertor





### Positive NSTX results motivate near-term plan with Upward LITER in NSTX-U

Confinement improvement attributed to reduced
 recycling

Cai, 2017 IEEE TPS

- Machines at PPPL showing benefits of lithium conditioning over *increasing areas*
  - TFTR Li "super-shots"
  - CDX-U, LTX
  - NSTX<sup>a</sup>
- Others examining Li:
  - FTU, T11-M, HT-7,EAST, DIII-D
  - TJ-II, RFX

<sup>a</sup>R. Maingi, et al., *Phys. Rev. Lett.* **107** (2011) 145004.







#### Liquid Metals – NSTX-U PAC-37 – Jan. 9th, 2018

### High-Z + Li NSTX-U would advance multiple Priority Research Directions simultaneously in unique ST facility

- NSTX and NSTX-U has emphasized surface science understanding of plasma-material interface (PRD-C,D)
  - MAPP deployed in 2016: year 1 capability
  - Studying impact of materials on plasma and vice-versa essential to whole-device modeling of material evolution
- A liquid metal-capable NSTX-U would enable high-Z vs. lithium comparison studies in a high-power device (PRD-A,B,C,E)
  - Extend LTX-beta studies
  - Builds on and complements continuing studies in higher A devices like EAST
- Heating power available to NSTX-U enables testing of liquid concepts to extreme power levels (PRD-A, B, C, E)
  - Potentially critical to ST line
  - Potentially critical to economical fusion energy concepts



### **One possible progression:** staged conversion to mitigate risk while enabling integrated performance assessment of the ST

- Full high-Z conversion enables comparison with conventional high-Z
- Provides high-power analog to evaluate LTX confinement results
- Open divertor and flexible magnetic configuration mitigates risk





#### The integrated assessment of the ST can be improved with near- and long-term PMI studies (Li + high-Z)

- Liquid metals have potential to solve several long-standing issues associated with solid PFCs
- Recent results continue to show confinement gains and potential for steady power exhaust
- US PMI Community identified the need to advance liquid metal science and technology, and evaluate feasibility
- A staged upgrade for NSTX-U can enable an integrated (plasma-material) assessment of the ST

- New PFCs require sustained R&D starting now for second 5-year plan





### Liquid metal PFCs can absorb more heat flux than leading tungsten technologies

- Actively cooled tungsten expected to survive 5-15 MW m<sup>-2</sup> steady-state
  - Transients and neutrons lead to ~5 MW m<sup>-2</sup>
  - Technological limitations lead to 9m major radius in EU DEMO1  $(P/R=17)^{a}$
- Tin CPS water-cooled target demonstrated 20 MW m<sup>-2</sup> in Magnum-PSI<sup>b</sup>
- Heat flux [MW/m<sup>2</sup> Unknown vapor-shielding limit
  - Dependent on replenishment rate to surface (technology)
  - **Dependent on core plasma** compatibility with divertor
- <sup>a</sup>Wenninger 2017 Nucl. Fusion; <sup>b</sup>Morgan, 2017 Nucl. Mater. Energy



**NSTX-U** 

#### Vapor shielding experiments in linear plasma device; shows path forward for vapor-box concept

- Liquid Sn and Li targets tested
  - Magnum-PSI linear plasma
  - Both exhibit temperature clamping
- Li-target tested based on pre-filled concept
- Vapor box 0D calculations indicate very high heat flux potential



 900
 DEMO reference pt.

 2.5GW fusion power

 200MW to outer div.

 700

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S. Van Eden, et al., *Nature Comm.* **8** (2017) 192. R. Goldston, et al., *Nucl. Mater. Energy* **12** (2017) 1118.

Rindt, PFMC 2017

1.0

#### Multiple tools developed to study complex, substratedependent lithium chemistry

- Quantum modeling by Krstic indicates preferential bonding of deuterium to oxygen in carbon matrix (theory)
- Laboratory studies by Capece show increased absorption by oxidized Li, but lower thermal decomposition temperature (laboratory studies)
- MAPP diagnostic successfully deployed in NSTX-U 2016 run campaign





#### MAPP Diagnostic (U-Illinois)

## Multiple decision points in the research program leading to new science and technology opportunities

- NSTX-U PFC Recovery project emphasis on prototypes builds confidence
- High-Z introduction enables early evaluation of operational impact of new PFC materials
  - LITER and U-LITER allows near-term examination of increased areal coverage with lithium conditioning
- Pre-filled target performance presents decision points:
  - Is a closed divertor upgrade necessary for high-temperature (i.e. vapor box)?
  - Is active cooling or fast-flow required for long-pulse particle flux handling?
- Closed loop facility enables broadened facility scope for research on integrated technologies





## Most mature technologies emphasized for current development path

- Bulk high-Z for high heat flux regions, coatings may be possible in low-heat flux regions
- Pre-filled targets build on graphite and high-Z substrate designs
- External Li feed into reservoir region with inertial cooling provides nearest target technology for NSTX-U



## Sustained effort needs to start now if this is to be implemented in 2025-2030 timeframe

- Reverse schedule developed to identify dependencies in R&D
  - Pre-filled high-heat flux tiles build on High-Z HHF
  - High-Z high-heat flux and low-heat flux tiles build on PFC Recovery activity
- Pre-filled tiles could be used in NSTX-U campaigns ~6 years from start (High-Z HHF in ~2 years)
- Li R&D facility recommended to accelerate development
  - Serial R&D processes indicate ~10 years of development to integrated PFCs
  - Fully flowing modules require Li loop and handling R&D & ops experience
  - Fully flowing modules required for Li facility that "runs as long as we want it to run"
  - Building/facility recommended in PPPL Li Corrective Action Plan to consolidate Li activities as a best practice



### A three-step progression can accelerate tests of flowing, liquid metal PFCs

1. High-Z divertor tiles + LITER

2. Pre-filled liquid-metal target

High Heat Flux, High-Z OED\_row2\_s **Divertor Upgrade** Conceptual design for pre-filled LM target e Flowing PFC loop integration C g 

3. Flowing LM PFC

**NSTX-U** 

High-Z divertor tiles + Li evaporated coatings provide divertor analogue of Magnum-PSI experiments

- 1. High-Z divertor tiles + LITER + U-LITER
  - Scientific goals:
    - Short-period particle-flux handling through temperature ramp
    - Quantify maintenance of Li on high-temperature substrate and protection of substrate
    - Re-examine suppression of erosion in high-flux divertor
    - Understand impact and core-edge compatibility of <u>high-temp. target</u> with limited inventory of Li





# Pre-filled targets test LM coverage, resupply and impact of significant Li source

- 2. Pre-filled liquid-metal target
  - Scientific goals:
    - Short-pulse particle-flux handling with *localized* Li source
    - Test models of maintenance of LM wetting and coverage
    - Understand limits of LM passive resupply
    - Understand impact and coreedge compatibility of <u>high-temp.</u> <u>target</u> with **larger** inventory of Li





### Final integration demonstrates LM introduction/extraction and inventory control

- 3. Flowing LM PFC
  - Scientific goals:
    - Short pulse particle flux handling with localized, constant material
    - Assess material inventory control to/from LM target
    - Understand performance of passive + active replenishment techniques
    - Understand impact and coreedge compatibility of <u>high-temp.</u> <u>target</u>





# Measurements of surface composition after lithium application indicate formation of stable oxide layer

Lithium/Oxygen Ratio - Vacuum, T<sub>s</sub> = 30 ± 10 °C



- Results shown from several lithium evaporation experiments
- Surface exclusively Li<sub>2</sub>O for ~100 hours at H<sub>2</sub>O pressure of 2 x 10<sup>-9</sup> T (upper figure)
- Persistence of maximum achievable plasma current – good plasma performance – suggests ability of Li<sub>2</sub>O to retain hydrogen (lower figure)
- Low resolution of MAPP XPS system means more time time needed for scans
  - Li<sub>2</sub>O formation from lithium unresolved for t < 1 hour</li>

#### Hydrogen retention in lithium and lithium oxide measured under controlled laboratory conditions

- Used stainless steel ultrahigh vacuum chamber with 2×10<sup>-10</sup> Torr base pressure
- Ni(110) single crystal high-Z substrate
- Lithium films created by thermal evaporation onto Ni substrate
  - Lithium oxide formed by O<sub>2</sub> exposure
- Lithium and lithium oxide films exposed to 500 eV H<sub>2</sub><sup>+</sup> ion beam for total total fluence of 4×10<sup>15</sup> H<sup>+</sup>cm<sup>-2</sup>
  - Beam defocused over surface to provide total fluence of 4×10<sup>15</sup> H<sup>+</sup>cm<sup>-2</sup>
- Hydrogen retention determined with temperature programmed desorption
  - Measurements made after irradiation with hydrogen at various lithium and lithium oxide film temperatures





## Experiments simulated with molecular dynamics calculations



### Classical molecular dynamics (MD) used

- Each atom treated as point mass
- Newton's equations integrated to compute motion
- Individual force equation for each atom derived from potential energy functional for system
- Amorphous target surfaces of pure Li and Li<sub>2</sub>O prepared for 90, 300, 400, 500, and 600K temperatures
- Computational cells of about 2000 atoms initially created at 300K for two cases
  - Random distribution of lithium atoms only
  - Predefined random distribution of 33% O and 67% of Li atoms

#### Amorphous distribution of lithium and lithium oxide in unit cells



- a) Li system (2000 lithium atoms)
- b) Li<sub>2</sub>O system: 33% of O (660 atoms) and 67% of Li (1340 atoms)
  - Green and red symbols represent Li and O, respectively

### Hydrogen retention in lithium and lithium oxide seen in laboratory experiments and MD simulations



- a) H retention fraction (H retention/H incident) in Li and Li<sub>2</sub>O as function of exposure temperature and comparison with D retention from MD calculations
- b) MD and experimental data on H retention fraction only in Li and Li<sub>2</sub>O as function of exposure temperature.

LTX demonstrated (again) confinement gains even with partial oxidation of Li coatings and Li pool

 Measured electron energy confinement exceeds neo-Alcator Linear Ohmic Confinement scaling ~3x (Boyle PRL 2017)



#### Liquid metal PFCs provide additional pathways for energy transport





### Liquid metal PFCs provide additional pathways for energy transport



- Conventional, solid PFCs utilize extrinsic impurities to enhance radiation
- Liquid metals access mobility as additional tool for energy transport
- Particle retention in Li a strong function of temperature links PMI physics to the technology

## An DEMO-relevant concept provides guidance for relevant technologies in NSTX-U

- Stability achieved by porous substrate
- Closely connected primary coolant and liquid lithium reservoir/supply structure
  - Temperature controls evaporation and condensation
  - Temperature controls hydrogen retention
- Continuous flow to/from the surface to flush gettered material and maintain wetted surfaces (substrate protection)

Jaworski, et al., Plasma Phys. Control. Fusion 55 (2013) 124040.



### Li Vapor Box (LVB) Divertor

- Li vapor divertor target is constrained in space by differential pumping
  - Bottom chamber reaches equilibrium density of ~ 10<sup>22</sup>/m<sup>3</sup> @ 750 Ċ
  - Upper chambers pump by condensation, at much lower temperature
- Plasma energy is dissipated in bottom box



## WallDYN: An integrated global model for mixed material migration



**WNSTX-U** Jacob H. Nichols – 59<sup>th</sup> APS-DPP 2017 – PP11.58 – Global Modeling of Wall Material Migration Following Boronization in NSTX-U 36

### WallDYN parameterizes plasma impurity transport with DIVIMP 2D Monte Carlo code

- 20k neutrals launched from each wall bin
  - Energy and angular distribution typical of sputtered particles
  - Deposition location, charge state recorded
- R ≡ Fraction of eroded flux from "source" that ends up in a certain charge state at "destination"
- 1 Algebraic equation for influx of each element & charge state:

$$\Gamma_{wall,elem}^{IN}(t) = \sum_{source} R_{source}^{wall} * \Gamma_{source,elem}^{OUT}(t)$$

- Assumptions:
  - Impurity concentrations are low enough not to disturb plasma ("trace impurity limit")
  - Plasma transport timescale << wall evolution timescale</li>
  - Plasma does not change in time



#### OSM/EIRENE edge plasma background generated for NSTX-U



- High strike point density/temperature, large radial gradients (H-Mode like)
- No external flow/pinch applied

### Results: agreement with OII emission trends greatly improved with thin film model



**NSTX-U** Jacob H. Nichols – 59th APS-DPP 2017 – PP11.58 – Global Modeling of Wall Material Migration Following Boronization in NSTX-U

### Results: agreement with BII, CII emission trends slightly improved with thin film model



- Thin film model brings WallDYN into good agreement with experimental oxygen emission trends following full boronizations
- No experimental trend in carbon emission following boronization
  - Good agreement with both WallDYN models
- No experimental trend in boron emission following boronization
  - Poor agreement with both WallDYN models
  - Reason for continued disagreement is under investigation

### Results: agreement with mini boronization emission trends similar with either model



- Thin film WallDYN model capable of differentiating between full and mini boronizations
  - Captures faster rise in impurity influx following mini boronization
- Both model and experiment show degradation of mini boronization coating on time scale of ~ 1 run day (15 sec plasma exposure)
- For mini boronization, thin film WallDYN model is nearly identical to old WallDYN model
  - Makes sense, because boron film applied to divertor (~30 A) is thinner than reaction layer (ion penetration depth ≈ 40 A)

### Results: surface concentration evolution during simulations



- Both WallDYN models underestimate the rise in oxygen surface composition observed in MAPP
- Surface boron agreement inconclusive
- Models do not take into account chemical state of surface
  - e.g. B-O vs. B-B vs. B-C bonds
  - May be important!

For more B/C/O surface discussion see: F. Bedoya JO4.10 Tue PM H. Schamis PP11.57 Wed PM

### Effect of more uniform boron coverage on oxygen evolution (*aka more GDC electrodes*)



 Uniformity of boron coatings is controlled by properties of He glow discharge and the number of anodes

### Effect of lower bulk oxygen concentration on oxygen evolution (*aka better bakeout*)



 Bulk oxygen content is typically controlled by pre-campaign conditioning (bakeout, etc.)